

# Synthesis and Characterization Of Some New Heterocyclic Compounds Such As: Oxadiazole and Azetidone-2-One Derivatives

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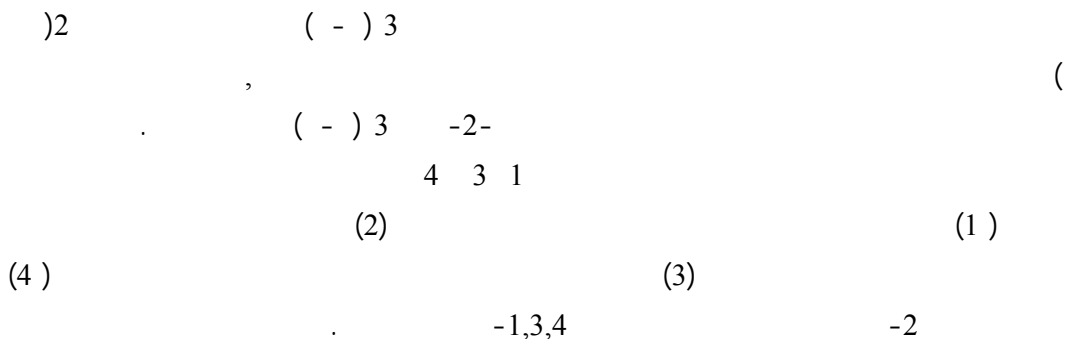
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## Abstract

A series of four membered ring the azetidone-2-one **3(a-c)** have been synthesized via Schiff bases **2(a-e)** with derivative of acetic acid in the presence of triethylamine with phosphorusoxychloride using dry methylene chloride under inert nitrogen atmosphere at 0 °C. to furnish the corresponding azetidone-2-one **3(a-c)** in moderate yields. on the other hand a series of five membered ring synthesized Substituted 1,3,4-oxadiazoles are of considerable pharmaceutical and material interest by treatment of a suspension of methyl salicylate hydrazide (**1**) in ethanol reflux with hydrazine to give 2-hydroxybenzohydrazide (**2**) this compound treated with Carbon disulfide (CS<sub>2</sub>) in the presence of sodium hydroxide to give 2-mercapto-1,3,4-Oxadiazole (**3**) this compound treated with hydrazine in ethanol to give 5-hydrazino-1,3,4-Oxadiazole (**4**). Finally treatment of (**4**) with 2-bromobenzaldehyde to gave 1,3,4—Oxadiazole derivative. The structural of these compounds were monitored and confirmed by using spectroscopic tools as IR and <sup>1</sup>HNMR spectra Elemental analysis.

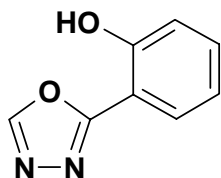


**Keywords:** medicinal chemistry, β-lactam, ketene, Staudinger reaction, four membered ring, 1,3,4-Oxadiazole

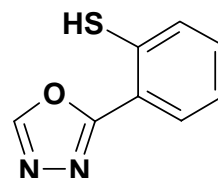
## Introduction

Substituted 1,3,4-oxadiazoles are of considerable pharmaceutical and material interest, which is documented by a steadily increasing number of publications and patents. For instance, 2-amino-1,3,4-oxadiazoles act as muscle relaxants<sup>1</sup> and show antibiotic activity.<sup>2</sup> Analgesic, antiinflammatory, anticonvulsive, diuretic and ant emetic properties are exhibited by 5-aryl-2-hydroxymethyl-1,3,4-oxadiazole derivatives,<sup>3</sup> and 2-hydroxyphenyl-1,3,4-oxadiazole acts as a hypnotic and as a sedative.<sup>4</sup> Some material applications of 1,3,4-oxadiazole derivatives lie in the fields of photosensitizers<sup>5</sup> and liquid crystals.<sup>6</sup> The common synthetic approaches to

oxadiazoles involve cyclization of diacylhydrazines<sup>7</sup>. A variety of reaction conditions influence the cyclization reaction. Typically, the reaction is promoted by heat and anhydrous reagents including thionyl chloride,<sup>8</sup> phosphorous oxychloride,<sup>9</sup> phosphorous pentoxide,<sup>10</sup> triphenylphosphine,<sup>11</sup> and triflic anhydride.<sup>12</sup> Alternative synthetic methods comprise the reaction of carboxylic hydrazides with ketenylidene triphenylphosphorane<sup>13</sup> or base-promoted cyclization reaction of trichloroacetic acid hydrazones.<sup>14</sup> Herein we report a simple method for the synthesis of 1,3,4-oxadiazoles having phenol<sup>15</sup> or thiophenol group like A and B compounds



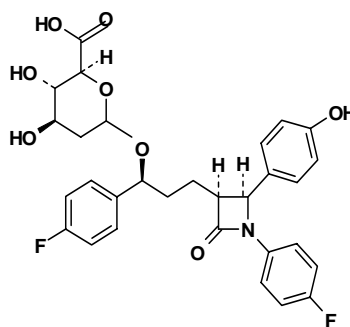
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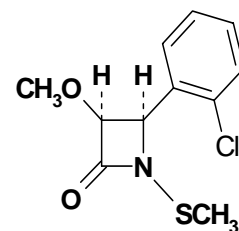
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$\beta$ -Lactam antibiotics still constitute one of the most widely employed class of antibacterial agents.<sup>16</sup> they continue to attract the attention of synthetic organic chemists as they present a variety of synthetic challenges. Besides this, they are useful as intermediates for  $\alpha$ - and  $\beta$ -amino acids, alkaloids, heterocycles, taxoids and other important compounds of biological

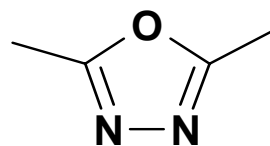
and medicinal interest.<sup>17</sup> In addition, recently some of the synthetic  $\beta$ -lactams have been reported to be biologically active as cholesterol acyl transferase inhibitors,<sup>18</sup> thrombin inhibitors,<sup>19</sup> human cytomegalovirus protease inhibitors,<sup>20</sup> matrix-metalloprotease inhibitors,<sup>21</sup> human leukocyte elastase,<sup>22</sup> and as cysteine protease<sup>23</sup>.



Hypcholesterolemic

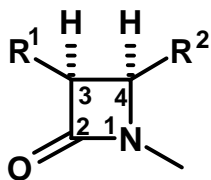


Antitumor



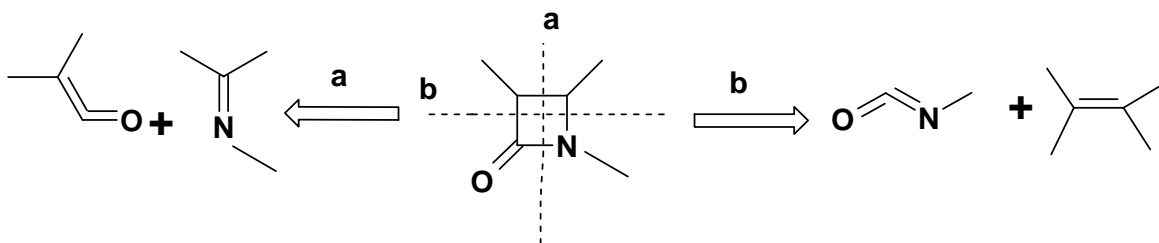
## Results and Discussion

Azetid-2-ones, commonly known as  $\beta$ -lactam constitute a well-known class of aliphatic heterocyclic compounds.



While the 1,3,4-Oxadiazole a well-known class of aromatic heterocyclic compounds

Two possible [2+2] cycloadditions can be envisaged for the synthesis of  $\beta$ -lactams (Scheme 1). One possibility consists of the [2+2] cycloaddition between ketenes and imines to yield  $\beta$ -lactams. This reaction has been explored experimentally and it is also known as the Staudinger reaction between ketenes and imines<sup>24-27</sup>. In an alternative approach, the [2+2] cycloaddition between alkenes and isocyanates leads to  $\beta$ -lactams. This reaction has been less extensively used, but it has proven to be useful in the chemical synthesis of interesting compounds **3(a-c)**.

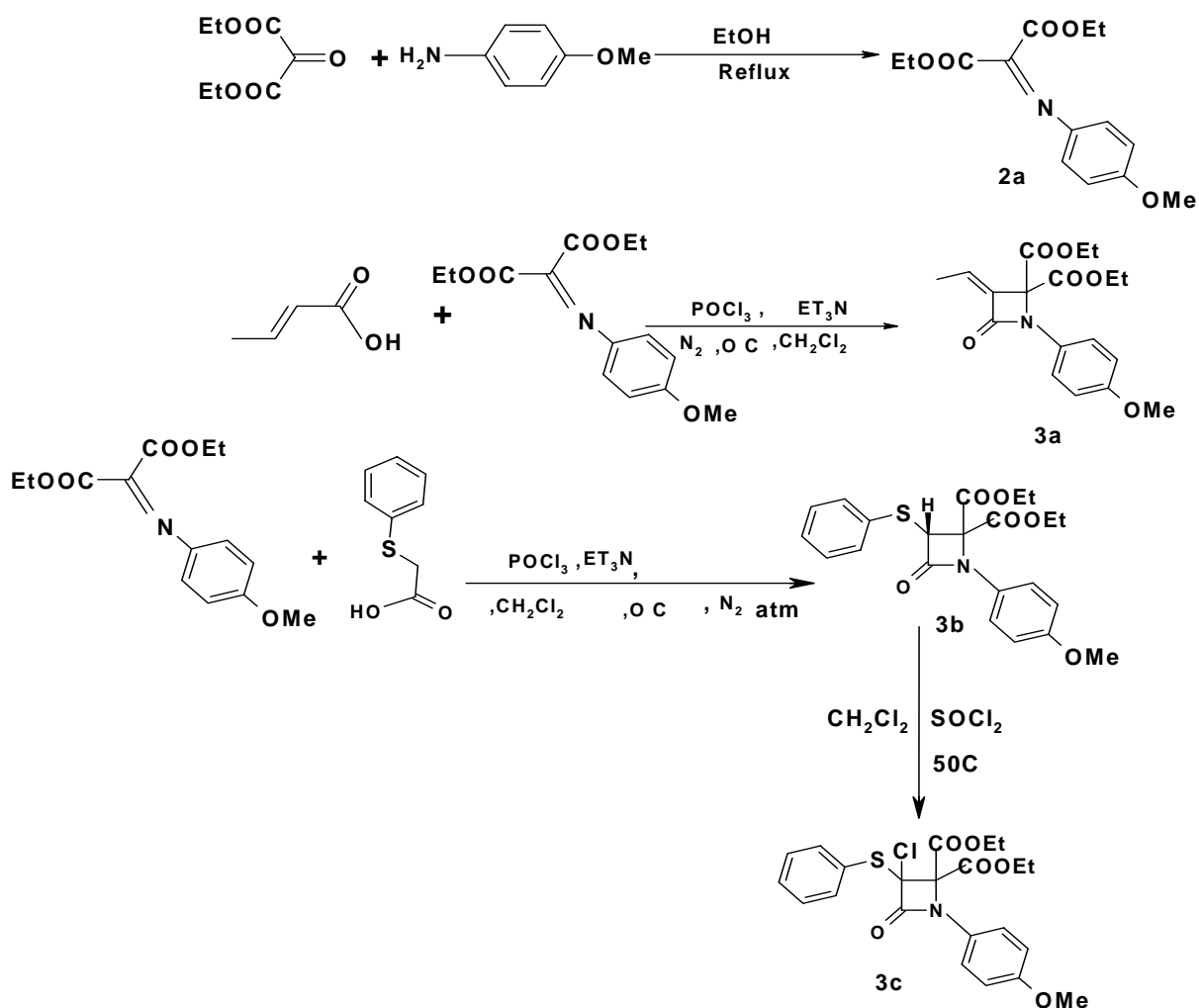


**Scheme 1: [2+2] disconnections of the  $\beta$ -lactam ring**

## FOUR MEMEBERD RING SYNTHESIS:

A series of 3-alkylidene 3-Phenylthio-3-Chloro azetid-2-one **3(a-c)** were

synthesized by the Staudinger ketene-imine [2 + 2] cycloaddition as shown below  
Scheme 2:



**Scheme2: The reaction between carboxylic acid with imine**

The IR spectra of the 3-alkylidene/3-phenylthio-3-chloroazetidine-2-one 3(a-c) were characterized by the presence of appeared the carbonyl group (amide carbonyl) at  $1758\text{cm}^{-1}$  and carbonyl ester at  $1740\text{cm}^{-1}$ , alkene and substituted ring which occurs within the ranges  $3085, 2980, 1600-1550$  and  $835-815\text{cm}^{-1}$ , respectively<sup>28</sup>.

The <sup>1</sup>H-NMR of 3-alkylidene β-lactam 3a showed four regions, an aliphatic region including one groups of signals at the region δ 1.2 ppm, corresponding to six equivalent protons  $2\text{CH}_3$  (6H, triplet, *J* coupling), second region showed methyl group at δ 2.2 (quartet) ppm, also methoxy group at 3.8 ppm (singlet). Finally region two equivalent methylene group at 4.3 ppm. ( $2\text{CH}_2$ , 4H, quartet, *J* coupling) The <sup>1</sup>H-NMR of 3-alkylidene β-lactam 3a

spectra showed one region, of the aromatic proton at (6.8-7.5ppm) corresponding to four aromatic protons (dd), Finally the proton at C3-H (-CH) showed at 6.1ppm and splitting to quartet are shown in (Figure (3-1)). The <sup>1</sup>H-NMR of 3-Phenylthio β-lactam 3b showed four regions, an aliphatic region including one groups of signals at δ 1.2ppm corresponding to two methyl group and signals at 4.3ppm corresponding to two methylene groups also signal group at δ 3.7ppm corresponding to methoxy group and signal peak at δ 5.3 ppm corresponding to C3-H. Finally the signals of aromatic protons at region δ 6.9-7.5 ppm corresponding to nine protons for two rings. are shown in Figure (3-2). The <sup>1</sup>H-NMR of 3-Phenylthio-3-chloro β-lactam 3c the signal of proton C3-H is absent and a

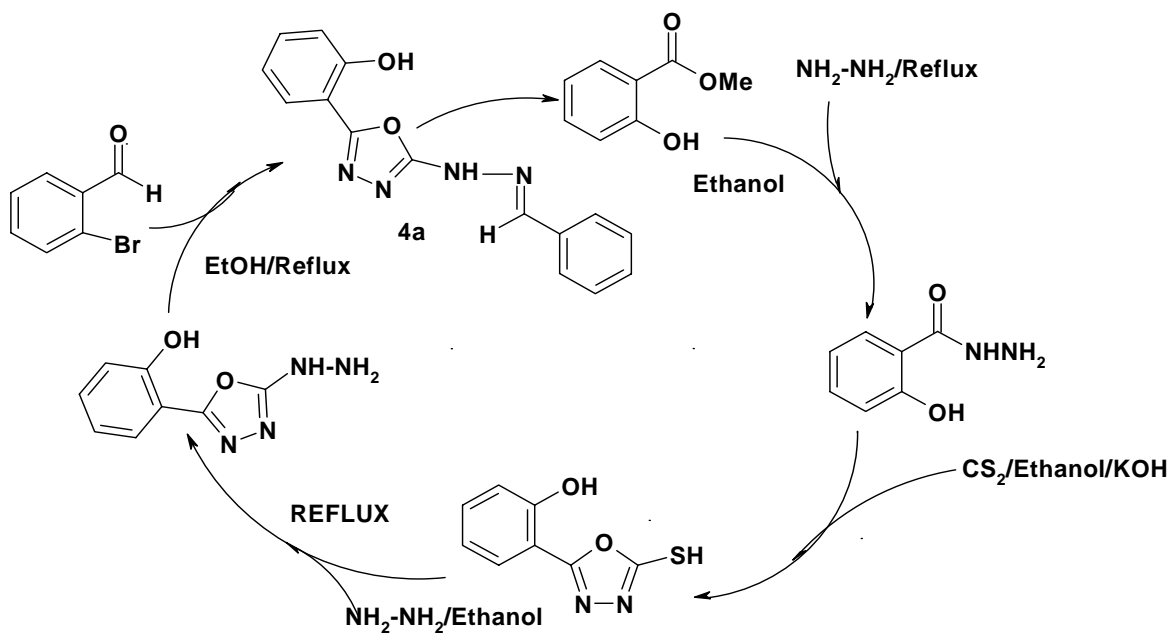
similar observations were found for 3b are shown in Figure (3-3).

The  $^{13}\text{C}$  NMR spectra 3c of the 3-Phenyl-3-chloro azetidine -2-ones showed The resonance at  $\delta$  163.55, 159.15 and 157.35 which were assigned to three carbonyl groups ,the signals of aromatic carbon at range  $\delta$ (113.9-136.4ppm) and the signals of aliphatic carbons ,the two methyl group at  $\delta$  13.9 ppm and two methylene at  $\delta$  63.37ppm.Finally the carbon of methoxy

group at  $\delta$  55.45ppm are shown in (Figure (3-4)).

### SYNTHESIS OF FIVE MEMEBERD RING:

In the present study2-Bromobenzaldehyde[1, 3, 4-oxadiazole -2-yl] hydrazone 5-(2-hydroxyphenol) 4a and some of their derivatives were synthesized .These synthetic reactions are summarized in Scheme3



SCHEME3

The IR spectra of 2-bromobenzaldehyde[5-(2-hydroxyphenyl)1,3,4-oxadiazole-2-yl]hydrazone methane 4a was characterized by the presence of phenolic and amine group which shows signals at  $3425\text{ cm}^{-1}$  and  $3392.5\text{ cm}^{-1}$  respectively as shown in Figure(3-5) and some signals at  $2933$ ,  $3033\text{ cm}^{-1}$  corresponding to aliphatic and aromatic carbon and also showed strong azomethane signal at  $1620\text{ cm}^{-1}$ . and some peaks at  $1348\text{ cm}^{-1}$ ,  $1431\text{ cm}^{-1}$  corresponding to stretching sym and asymmetry of C-O-C

The  $^1\text{H-NMR}$  of benzaldehyde(5-(2-hydroxyphenyl)1,3,4-oxadiazole-2-yl)hydrazone methane 4a showed two regions, the aromatic protons give multiplets signals at 6.87-8.87ppm with azomethine proton (CH-N)<sup>29</sup>.Also showed broad singlet at 10.3ppm corresponding to hydrogen of hydroxyl group bonding with the nitrogen atom

### Experimental

All melting points are uncorrected and are expressed in degree( $^{\circ}\text{C}$ ), using melting point SMP<sub>3</sub>. IR spectra were recorded asKBr disks using shimadzu FT-IR 8400 using KBr disks.  $^1\text{H}$  NMR

spectra were recorded using Bruker system AL 300 (300 MHz) and tetramethylsilane (TMS) as internal standard.  $^{13}\text{C}$  NMR spectra were recorded using Bruker system AL 300 (300 MHz) and tetramethylsilane (TMS) as internal standard  $^{13}\text{C}$  NMR.

### General procedure of imine synthesis 2a

#### N-(4-Methoxyphenyl)-1,1-diethoxycarbonylimine 2a

A mixture of diethyl ketomalonate (0.7 g, 1 mmol) and *p*-anisidine (0.5 g, 1 mmol) was refluxed in dry benzene on a heating mantle using a Dean-Stark apparatus. The reaction was monitored by TLC. After 4-5 h, when there was no spot left corresponding to the starting materials, benzene was removed under reduced pressure and the crude product 2a (1.0 g, 94%) thus obtained, as a liquid was used as such for the subsequent reactions. It showed following spectral data: IR ( $\text{CHCl}_3$ ): 1675, 1630, 1515, 1510  $\text{cm}^{-1}$ ;  $^1\text{H-NMR}$  ( $\text{CDCl}_3$ )  $\delta$ : 1.4 (q, 3H,  $\text{COOCH}_2\text{CH}_3$ ), 1.5 (t, 3H,  $\text{COOCH}_2\text{CH}_3$ ), 3.8 (s, 3H,  $\text{OCH}_3$ ), 4.25 (q, 2H,  $\text{COOCH}_2\text{CH}_3$ ), 4.45 (q, 2H,  $\text{COOCH}_2\text{CH}_3$ ), 6.97 (dd, 4H, AB pattern, aromatic protons).

### General procedure of azetidine-2-one

#### 3(a-c)

#### N-(4-Methoxyphenyl)-3-ethylidene-4,4-diethoxycarbonyl azetidin-2-one 3a

To solution of 2-butenic acid (1a) (0.45 g, 1.5 mmol), imine 2a (0.5 g, 1 mmol) and triethylamine (0.54 g, 3 mmol, 0.75 ml) in 80 mL dry methylene chloride was added dropwise under nitrogen atmosphere at  $0^\circ\text{C}$ , a solution of phosphorus oxychloride ( $\text{POCl}_3$ ) (0.41 g, 0.24 mL, 1.5 mmol) in 20 mL of dry methylene chloride with constant stirring. The reactant was stirred overnight at room temperature. The completion of reaction was monitored by TLC. After the completion, the contents were washed successively with 1N HCl

(30 ml), water (3x30 ml, 5%  $\text{NaHCO}_3$  (30 ml) and brine (30 mL). The organic layer was separated and dried over anhydrous  $\text{Na}_2\text{SO}_4$ . The solvent was removed under reduced pressure and the crude product was purified by column chromatography using silica gel eluting with 10% ethyl acetate : hexanes. Solvent evaporation furnished pure  $\square$ -lactam 3a (0.66 g, 60%). Its structure was confirmed on the basis of following spectral data: m.p.:  $90-92^\circ\text{C}$ ; IR ( $\text{CHCl}_3$ ): 1758, 1740  $\text{cm}^{-1}$ ;  $^1\text{H-NMR}$  ( $\text{CDCl}_3$ )  $\delta$ : 1.2 (t, 6H,  $J = 7$  Hz,  $2 \times \text{COOCH}_2\text{CH}_3$ ), 2.2 (q, 3H,  $\text{CH}_3$ ), 3.77 (s, 3H,  $\text{OCH}_3$ ), 4.40 (q, 4H,  $J = 7$  Hz,  $2 \times \text{COOCH}_2\text{CH}_3$ ), 6.10 (q, 1H,  $\text{C}_3\text{-H}$ ), 6.8-7.5 (dd, 4H, aromatic protons)

#### 1-(4'-Methoxyphenyl)-3-phenylthio-4,4-diethoxycarbonylazetidin-2-one 3b

To solution of phenylthioacetic acid (0.45 g, 1.5 mmol), imine 2a (0.5 g, 1 mmol) and triethylamine (0.54 g, 3 mmol, 0.75 mL) in 80 mL dry methylene chloride was added dropwise under nitrogen atmosphere at  $0^\circ\text{C}$ , a solution of phosphorus oxychloride ( $\text{POCl}_3$ ) (0.41 g, 0.24 mL, 1.5 mmol) in 20 mL of dry methylene chloride with constant stirring. The reactant were stirred overnight at room temperature. The completion of reaction was monitored by TLC. After the completion, the contents were washed successively with 1N HCl (30 mL), water (3x30 mL), 5%  $\text{NaHCO}_3$  (30 mL) and brine (30 mL). The organic layer was separated and dried over anhydrous  $\text{Na}_2\text{SO}_4$ . The solvent was removed under reduced pressure and the crude product was purified by column chromatography using silica gel eluting with 10% ethyl acetate : hexanes. Solvent evaporation furnished pure  $\square$ -lactam 3b (0.66 g, 60%). Its structure was confirmed on the basis of following spectral data: m.p.:  $94-95^\circ\text{C}$ ; IR ( $\text{CHCl}_3$ ): 1758, 1740  $\text{cm}^{-1}$ ;  $^1\text{H-NMR}$  ( $\text{CDCl}_3$ )  $\delta$ : 1.2 (t, 6H,  $J = 7$  Hz,  $2 \times \text{COOCH}_2\text{CH}_3$ ), 3.77 (s, 3H,  $\text{OCH}_3$ ), 4.40 (q, 4H,  $J = 7$  Hz,  $2 \times \text{COOCH}_2\text{CH}_3$ ), 5.20 (s, 1H,  $\text{C}_3\text{-H}$ ), 6.8-

7.5 (m, 9H, aromatic protons);  $^{13}\text{C-NMR}$  ( $\text{CDCl}_3$ )  $\delta$  : 13.8, 14.16, 55.20, 62.11, 62.76, 72.4, 113.9, 121.4, 127.3, 129.32, 129.5, 130.3, 133.9, 157.3, 161.3, 165.5, 165.8; Anal. Calcd. for  $\text{C}_{22}\text{H}_{23}\text{O}_6\text{NS}$  : C, 61.53; H, 5.36; N, 3.26; Found: C, 61.40; H, 5.29; N, 3.21.

### 1-(4-Methoxyphenyl)-3-Chloro-3-phenylthio-4,4-diethoxycarbonylazetid-2-one (3c)

To a well stirred solution of  $\alpha$ - phenylthio- $\beta$ -lactam 3b( 0.9g, 2mmoles )in 50 ml dry methylene Chloride , under nitrogen at  $0^\circ\text{C}$  , was added a solution of sulfuryl Chloride ( $\text{SO}_2\text{Cl}_2$ ) ( 0.39g , 2mmol,0.2ml ) in 10 ml dry methylene Chloride in 10 minutes contents were stirred for additional half hour . The progress of reaction was monitored by TLC. Solvent evaporation followed by column chromatography on silica gel using ethylacetate : hexanes(1:10) yielded pure  $\beta$ - lactam 3c (1.0g , 75%) ,IR:  $1760,1720\text{ cm}^{-1}$  ,  $^1\text{HNMR}$  ( $\text{CDCl}_3$ )  $\delta$ 1.26 (t,6H,  $2\text{XCOOCH}_2\text{CH}_3$ ),3.78(s,3H, $\text{OCH}_3$ ), 4.35(q,4H,  $2\text{XCOOCH}_2\text{CH}_3$ ),6.83-7.73 (m,9H,aromatic protons) .  $^{13}\text{CNMR}$ ( $\text{CDCl}_3$ )  $\delta$ :13.94, 29.70,55.459 , 63.37 , 113.97, 120.99 121.32 ,127.36, 128.96, 129.15, 129.32,130.31, 130.47, 136.42 , 157.35, 159.15, 163.55

### General procedure for synthesis of 2-bromobenzaldehyde[5-(2-hydroxyphenyl) 1,3,4-oxadiazole-2-yl] hydrazone methane 4a and some of their derivative

#### 1-Synthesis of 2-hydroxybenzohydrazine

To mixture of 0.1mole of methylsalicylate with 0.2mole of hydrazine hydrate in 100 ml of absolute ethanol .The mixture was thoroughly stirred and heated under reflux for 5h,the reaction time was monitored through TLC technique after completion of reaction ,the solution was concentrated to small volume and the residue was dissolved in water, this furnished precipitate which was filtered, washed and recrystallized from aqueous ethanol Yield %95, m.p =148-150  $^\circ\text{C}$

#### 2-Synthesis of 2-hydroxybenzohydrazine 2-(5-mercapto-1,3,4-Oxadiazole-2-yl)phenol

To mixture of 0.1mole of hydrazide with 0.1mole of KOH in 100 ml of absolute ethanol and add 0.1mole of carbon disulfide. The mixture was thoroughly stirred and heated under reflux for 5h,the reaction time was monitored through TLC technique after completion of reaction ,the solution was concentrated to a small volume and the residue was dissolved in water,this solution was acidified to pH 2-3 by addition of dil.hydrochloric acid and this furnished a precipitate which was filtered,washed and recrystallized from aqueous ethanol Yield %70, m.p = (203-205)  $^\circ\text{C}$

#### 3) Synthesis of 2-(5-hydrazino-1,3,4-Oxadiazole-2-yl)phenol

To mixture of 0.1mole of 2-(5-mercapto-1,3,4-Oxadiazole-2-yl)phenol with 0.1mole of hydrazine hydrate in 100mL of absolute ethanol .The mixture was thoroughly stirred and heated under reflux for 5h,the reaction time was monitored through TLC technique after completion of reaction ,the solution was concentrated to a small volume and the residue was dissolved in water and this furnished a precipitate which was filtered, washed and recrystallized from aqueous ethanol. Yield %70, m.p= 203-205  $^\circ\text{C}$

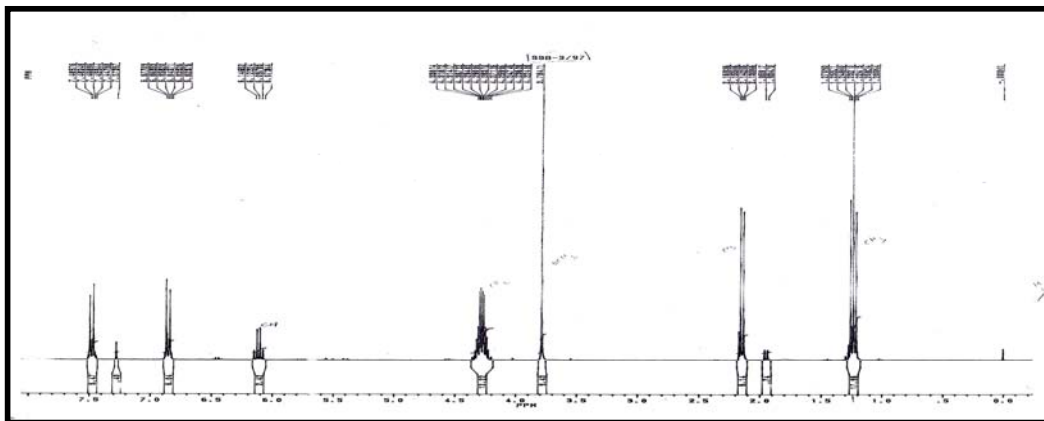
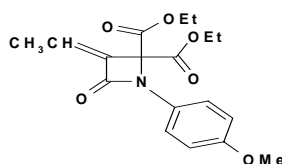
#### 4) Synthesis of Benzyldehyde(5-(2-bromophenyl)-1,3,4-oxadiazole -2-yl)hydrazone-methane 4a

To mixture of 0.02 mole of 2-hydroxybenzohydrazine2-(5-hydrazino-1,3,4-Oxadiazole-2-yl)phenol with 0.02mole of 2-bromobenzaldehyde in 50mL of absolute ethanol .The mixture was thoroughly stirred and heated under reflux for 3h,the reaction time was monitored through TLC technique after completion of reaction ,the solution was concentrated to a small volume and the residue was dissolved in water and this furnished a precipitate which was filtered, washed and recrystallized from aqueous ethanol. Yield % 62.6, m.p = 240-238  $^\circ\text{C}$

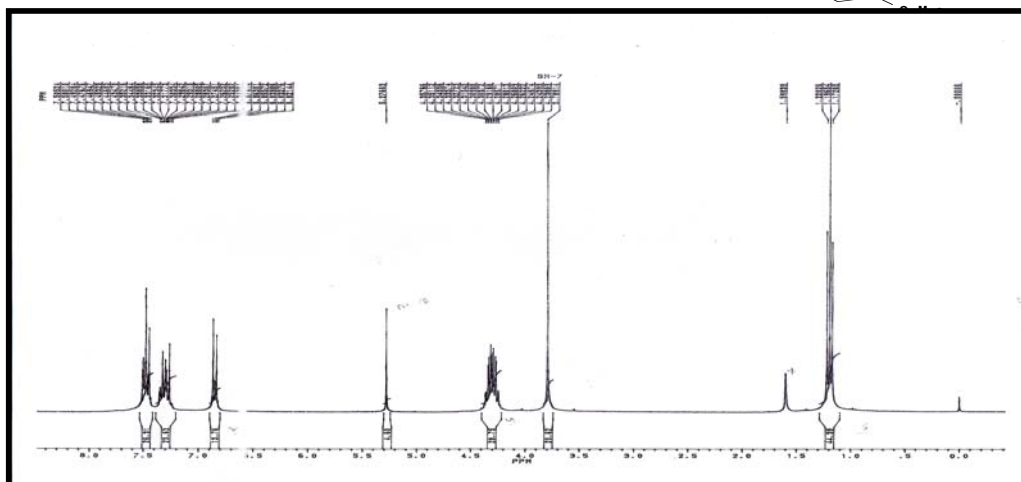
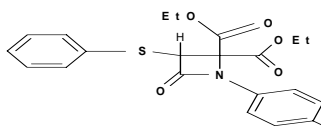
### Acknowledgments

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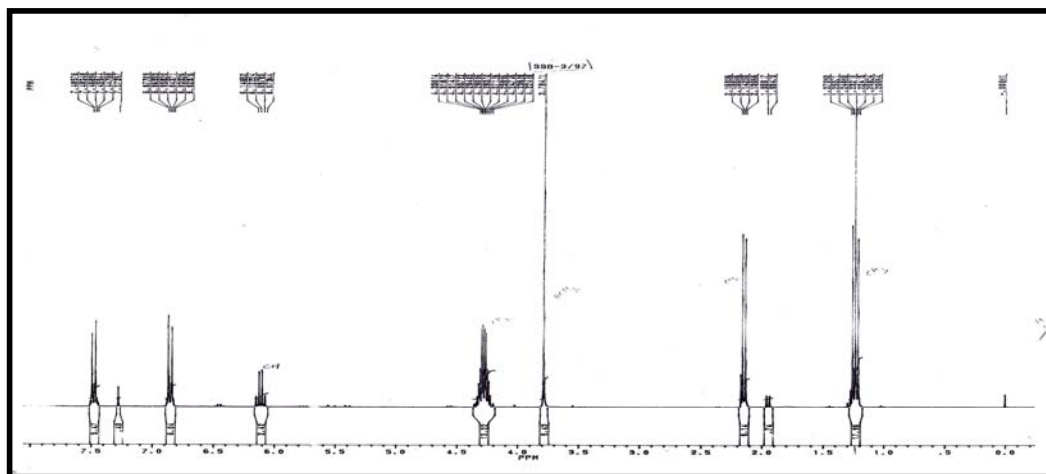
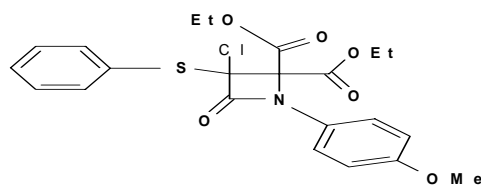


Figure(3-1)  $^1\text{H-NMR}$  of N-(4'-Methoxyphenyl)-3-ethylidene-4,4-diethoxycarbonyl azetidin-2-one (3a)

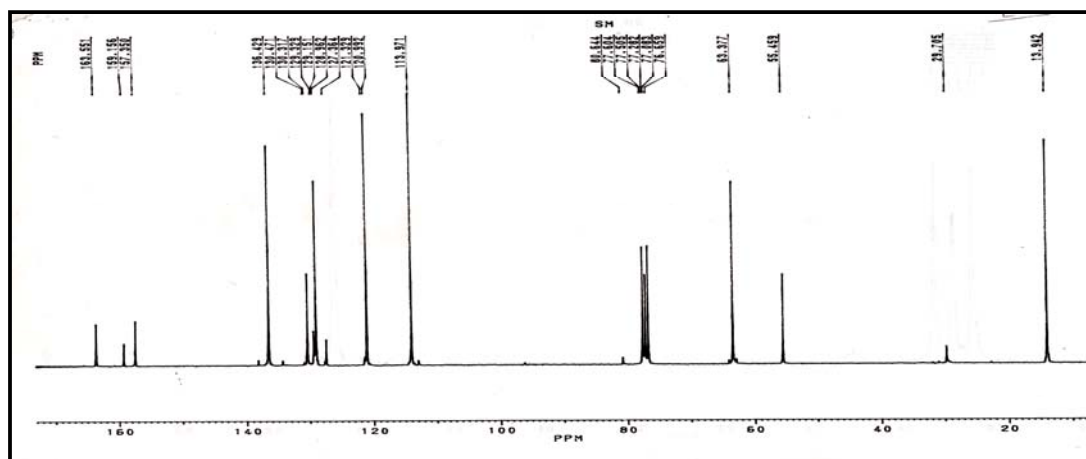
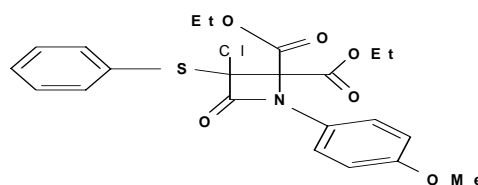


Figure(3-2)  $^1\text{H-NMR}$  of N-(4'-Methoxyphenyl)-3-phenylthio-4,4-diethoxycarbonyl azetidin-2-one 3b

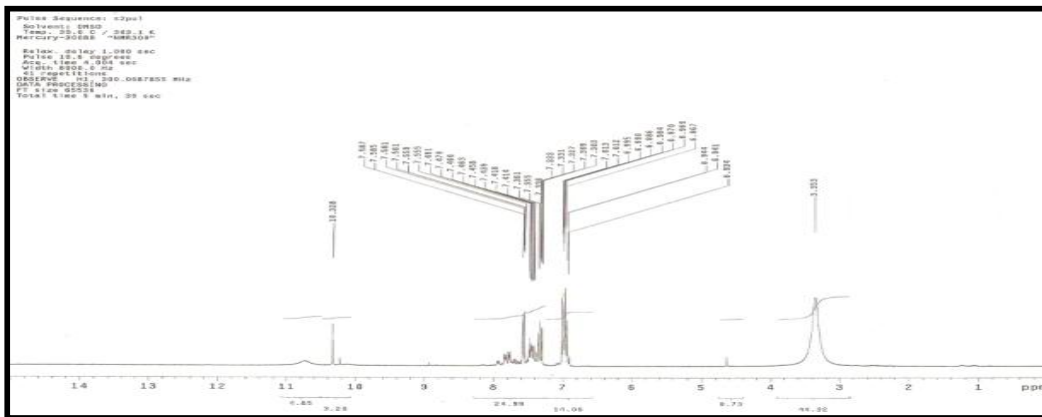
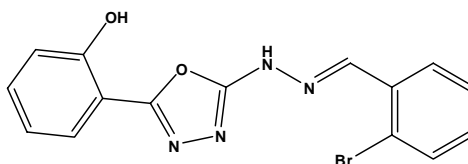




Figure(3-3)  $^1\text{H-NMR}$  of N-(4'-Methoxyphenyl)-3-phenylthio-3-chloro-4,4-diethoxycarbonyl azetidin-2-one 3c



Figure(3-4)  $^{13}\text{C-NMR}$  of N-(4'-Methoxyphenyl)-3-phenylthio-3-chloro-4,4-diethoxycarbonyl azetidin-2-one 3c



Figure(3-4)  $^1\text{H-NMR}$  of Benzyldehyde(5-(2-bromophenyl)-1,3,4-oxadiazole -2-yl)hydrazone-methane 4a

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